



Development of environment Friendly Clean Agent for Replacement of Halons Used In Combat Tanks.

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Abstract

The present work is an effort to control the depletion of ozone layer by replacement of halons fire extinguishant used in combat vehicles with suitable environment friendly near equivalent clean alternatives. In this context other fire extinguishing agents like fluorocarbons having less ozone depletion potential have been experimented by using actual hazard volume of combat Tanks and observed that the effectiveness and capability of fluorocarbon fire extinguishing agents are near

Keywords: Halons, fire extinguishing systems, Hexafluorocarbon, Aqueous Film Forming Foam, Chlorofluorocarbons, Combat vehicles, Ozone depletion potential, Global warming potential, Montreal protocol, Fire detection and suppression system.

1. Introduction

Halon 1301 has been used for decades as the primary fire extinguishing material for a multitude of military applications. However, Halons have very high ozone depleting potentials which results in higher levels of ultraviolet radiation at Earth's surface and gives rise to serious health effects therefore its production was stopped in 1994 in most of the world. As per Montreal protocol use of Halons have been banned however developing nations are allowed to use Halons up to 2010 which has been further extended temporarily for its mission critical applications. Accordingly research initiated to identify and develop replacement agents and technologies to satisfy the performance requirements of fire protection in combat vehicles.

Halons are used in crew compartment, hand held extinguishers and engine compartment of Tanks Accordingly this research will be based

on Halon elimination efforts in three separate grounds of combat vehicle applications. The research program is to identify alternatives to Halons used in fire extinguishing systems (FES) of Army ground based Combat tanks and trucks .

Based on the requirements, individual chemical agents having near equivalent fire extinguishing capability with low ozone depleting potentials have been experimented.

Initial investigations indicated that a universal solution would not be available for drop in replacement of Halons Accordingly it is decided to develop near equivalent clean agent having similar fire extinguishing property.

2. Synopsis

The research has been divided in three stages which are given as under-

STAGE - I Comprehensive study of near



equivalent fire extinguishing agents.

The study was based on a review of the extensive research and engineering literature covering the physical and chemical processes active in flames and involved in flame extinguishment. Published lists of prospective Halon replacements have been evaluated. Ozone depletion potential (ODP), a useful metric found in regulatory legislation, has been examined in the light of recent work for alternative agents. Also Global warming potential (GWP), a measure of agent effect on climate, has been examined

STAGE -II Testing for fire extinguishing capability of agents for effectiveness.

The extinguishing concentrations of gaseous agents are determined by small-scale tests. Developmental testing of the most promising concepts, by simulating hazard area of existing combat vehicles. Crew Survivability Criteria has been taken into consideration for the minimum acceptable requirements of automatic fire extinguishing systems for occupied vehicle compartments.

STAGE -III Technical feasibility of agents wrt existing technology in Combat vehicles-

Technical feasibility of agents wrt existing fire extinguishing systems and technology in combat vehicles, test methodology, including agent toxicology, storage stability, and extinguishment effectiveness, have been studied. The full-scale test simulating actual hazard area of existing combat vehicles, thermal decomposition products and potential for retro fitment wrt most promising alternatives on existing platforms to be examined.

3. Methodology of Experiments-

This research for halon replacements has been based on modifying the molecular structure of halons to reduce or eliminate the chlorine and bromine atoms which are responsible for ozone depletion. In this connection halon-like halocarbons includes hydrofluorocarbons (HFCs), hydrochlorofluorocarbons (HCFCs), Viz- FE-13, FE-25, FM-200, FE-36 have been studied and most promising agents have been tested for fire suppression performance. The extinguishing concentrations of gaseous agents are determined by small-scale tests which are given as under

| Agents Trade Name | Chemical Formula | Chemical Name | Vapor Pressure | Boiling Point (F) | Ozone Depletion Potential (ODP) | Liquid Density @77 F(lb/ft ³) | Performance factor(Hept) CB Ext Conc % |
|-------------------|------------------|---------------------------------|----------------|-------------------|---------------------------------|---|--|
| FM-200 | C3F7H | 1,1,1,2,3,3,3heptafluoropropane | 66.5 | 2.5 | 0 | 86.7 | 5.8 |
| FE-36 | C3F6H2 | 1,1,1,3,3,3 hexafluoropropane | 39.9 | 33.2 | 0 | 85.5 | 5.6 |
| FE-13 | CF3H | Trifluoromethane | 665 | -115.7 | 0 | 41.8 | 12 |
| FE-25 | C2F5H | Pentafluoroethane | 190 | -55.3 | 0 | 78.0 | 8.1 |
| Halon 1301 | CF3Br | Bromo Trifluoromethane | 234.8 | -72.0 | 12- 16 | 96.0 | 2.9 |

4. Conclusion

We are aggressively pursuing alternatives to Halon 1301 in ground combat vehicles. As of now, we are reliant on Halons for fire detection and suppression system of our

combat Tanks, but the use of Halons has been banned as per Montreal protocol. Developing nations are allowed to use Halons in mission critical application up to 2010. Accordingly Halocarbon agents having similar molecular structures to halon are



modified to reduce or eliminate the chlorine and bromine atoms that are responsible for ozone depletion for all combat vehicle applications have been identified and experimented Their fire extinguishing performance has been evaluated It has been

shown that the agents are able to extinguish fires at their design concentration. However most agents listed above having higher extinguishing concentrations than Halon 1301 and, therefore, are less effective.

5. References

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