

Synthesis, Spectral Studies and Biological Screening of Novel Series of Chalcones

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Abstract

A novel series of chalcones (3a-k) have been synthesized by Claisen-Schmidt condensation between aromatic ketone and several aromatic aldehydes in the presence of potassium hydroxide. The structures of the compounds were confirmed by FTIR, ¹HNMR, MASS spectral data. All the novel chalcones were screened for their antimicrobial activities(agar disc-diffusion method).

Keywords: Chalcone, Spectral Studies, Biological activity.

1. Introduction

True importance of chalcone is extended into two branches, One is biological activity associated with them which includes antimalarial (Motta ,2006; Awasthi et al, Med Chem Res, 2009 and Lim SS, 2007), anticancer (Achanta G .2006; Echeverria C, 2009 and Ilango K, 2010), anti-inflammatory (Yadav HL,2010 and Zhang XW,2010), antimicrobial (Hamdi N,2010 ; Bhatia NM,2009; Awasthi SK, Amer J Top Med Hyg, 2009; Bag S,2009 and Lahtchev KL,2008) anticonvulsant (Kaushik S.2010), antioxidant (Vasil'ev RF, 2009; Sivakumar PM ,2010 and Vogel S,2008), xanthine oxidase inhibitor, aldol reductase inhibitor, epoxide hydrolase inhibitor (Najafian M, 2010; Zarghi A ,2006 and Chimenti F ,2009), and other one is its use as a template for synthesis of various heterocyclic compounds such as pyrimidine, pyrazoline, benzofurans, thiadiazines. isoxazole, quinolinones, benzodiazepine, etc. Some of these synthesized compounds show significant therapeutic activity (Gaede BJ,1993; Shibatai K,1993; EI- Hamouly Ws ,2011and Abonic R,2008). There are different methods reported for the synthesis of chalcone

analogues, most commonly used one is Claisen-Schmidt condensation reaction. They can be synthesized conventionally by acid or base catalyzed method. However, Green chemistry approaches are more beneficiary to synthesize chalcone derivatives so as to avoid pollution, use of toxic reagents and strongly acidic or basic condition (Jayapal MR,2010). The different traditional methods used for synthesizing these chalcones such as base catalyzed (NaOH, KOH, Ba(OH)₂) and acid catalyzed (including Lewis acids) condensation processes in the presence of suitable solvent, many new eco-friendly methods like use of ultrasonic radiations (Chtourou, M,2009), microwave assisted (Abdel-Aziz, H. A, 2011), solvent free synthesis by grinding (Rateb, N. M, 2005).

2. Materials And Methods

Experimental

Melting points were uncorrected and determined in open capillaries. The purity of the compound is checked by TLC. The IR spectra were recorded on Shimadzu FT-IR Spectrometer using potassium bromide pellets,

Abbas Biradar, P.A. Kulkarni

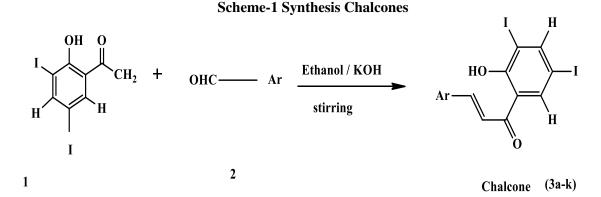


and ¹HNMR spectra were recorded on a Bruker Avance II 400 Spectrometer against TMS as internal standard. Mass spectra were recorded on Waters Micromass Q-Tof Micro spectrometry.

General Procedure for the Synthesis Chalcones

A mixture of substituted benzaldehyde (1

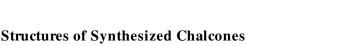
mmol) with 2hydroxy, 3,5-diiodo acetophenone(1 mmol) in ethanol (40 ml) was added (2 mmol) solution of potassium hydroxide. The reaction mixture was then stirred for 1 hr in ice bath, after completion of the reaction (monitored by TLC) the reaction mixture was poured into an ice cold solution of water. The solid obtained washed with and crystallized from water ethanol.

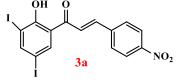


3. Result and Discussion

We have reported the newly chalcones which were carried out according to the Claisen-Schmidt condensation of an ortho hydroxy ketone with several aromatic aldehyde as indicated to Scheme 1.

The synthesized chalcones confirmed by IR spectral data showing sharp bands in the range between $1600 - 1660 \text{ cm}^{-1}$ indicated the presence of the C=O group. Chalcones (3a-k) were also confirmed by ¹HNMR spectral analysis. Inspection of the ¹HNMR spectra



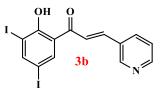


(E)-1-(2-hydroxy-3,5-diiodophenyl)-3-(4nitrophenyl)prop-2-en-1-one

suggested that the chalcones were geometrically pure and configured trans $(JH_{\alpha}=H_{\beta}=15Hz)$.

The antimicrobial activity of the newly chal cones was evaluated by

Disc diffusion Method (H. Afaf, 2000)

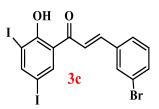


(*E*)-1-(2-hydroxy-3,5-diiodophenyl)-3-(pyridin-3-yl)prop-2-en-1-one

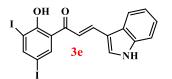


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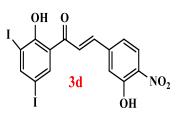
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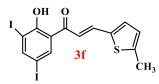
(*E*)-3-(3-bromophenyl)-1-(2-hydroxy-3,5diiodophenyl)prop-2-en-1-one



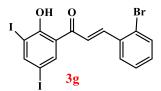
 $(E)\mbox{-}1\mbox{-}(2\mbox{-}hydroxy\mbox{-}3,\mbox{-}diiodophenyl)\mbox{-}3\mbox{-}(1H\mbox{-}indol\mbox{-}3\mbox{-}yl) \prop\mbox{-}2\mbox{-}en\mbox{-}1\mbox{-}one$



(E)-1-(2-hydroxy-3,5-diiodophenyl)-3-(3-hydroxy-4nitrophenyl)prop-2-en-1-one

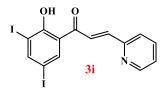


 $(E) \mbox{-}1\mbox{-}(2\mbox{-}hydroxy\mbox{-}3,5\mbox{-}diiodophenyl)\mbox{-}3\mbox{-}(5\mbox{-}methylthiophen\mbox{-}2\mbox{-}yl) \mbox{prop-}2\mbox{-}en\mbox{-}1\mbox{-}one$

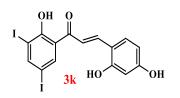




(*E*)-3-(2-bromophenyl)-1-(2-hydroxy-3,5-diiodophenyl) prop-2-en-1-one



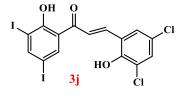
(*E*)-1-(2-hydroxy-3,5-diiodophenyl)-3-(pyridin-2-yl)prop-2en-1-one



(E)-3-(2,4-dihydroxyphenyl)-1-(2-hydroxy-3,5-diiodophenyl)p rop-2-en-1-one

prop-2-en-1-one

(E)-1-(2-hydroxy-3,5-diiodophenyl)-3-(pyridin-4-yl)



(*E*)-3-(3,5-dichloro-2-hydroxyphenyl)-1-(2-hydroxy-3,5diiodophenyl)prop-2-en-1-one



Table-1 The Physical data of Synthesized Chalcones(3a-k)							
Entry	Product	Mol. Formula	Yield %	M.P. (° C)			
1	3a	$C_{15}H_9O_4NI_2$	90	194			
2	3b	$C_{14}H_9O_2NI_2$	85	192			
3	3c	$C_{15}H_9O_2I_2Br$	80	162			
4	3d	$C_{15}H_9O_5NI_2$	90	110			
5	3e	$C_{17}H_{11}O_2NI_2$	92	150			
6	3f	$C_{14}H_{10}O_2 I_2S$	94	108			
7	3g	$C_{15}H_9O_2I_2Br$	80	130			
8	3h	$C_{14}H_9O_2NI_2$	80	174			
9	3i	$C_{14}H_9O_2NI_2$	75	100			
10	3ј	$C_{15}H_8O_3I_2Cl_2$	75	166			
11	3k	$C_{15}H_{10}O_4I_2$	90	170			

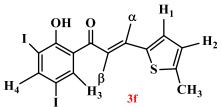
Spectral analysis of selected chalcones:



FTIR (**KBr**, **cm**⁻¹):- 1640 (C=O) ,1589(C=C) , 1427 (C-C Aromatic str) . ¹**HNMR :-** 7.19(d ,1H₁), 7.62(d ,1H₂), 7.66(d ,1H₃), 7.78(d ,1H₄), 7.81(d ,1H ,Hα, J=15Hz), 8.17(d , 1H , Hβ,J=15Hz), 8.17(s , 1H , H₅),

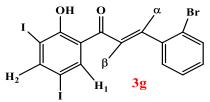
8.22(s,1H, H₆), 13 (s,1H,OH).

M.S. (M/z):- 521(M+1).



FTIR (KBr, cm⁻¹):- 1630 (C=O) ,1557(C=C) , 1433(C-C Aromatic str).

¹**HNMR :-** 2.52(s, 3H, CH₃), 6,69(d ,1H₁), 6.91(d ,1H₂), 7.55(d ,1H ,Hα, J=15Hz), 7.98(d, 1H, Hβ,J=15Hz), 8.18(d ,1H₃), 8.23(d ,1H₄), 8.17(s ,1H, H₅), 8.22(s ,1H, H₆), 13.90 (s,1H,OH). **M.S. (M/z):-** 521(M+1).



FTIR (**KBr**, **cm**⁻¹):- 1640 (C=O) ,1588(C=C) ,1429(C-C Aromatic str), 665(C-Br). ¹**HNMR :-** 7.67 (d ,1H ,Hα, J=15Hz), 7.35-8.02(m ,5H H-Aromatic), 8.20(d ,1H, H₅), 8.24(d, 1H, Hβ,J=15Hz), 8.53(s ,1H, H₆), 13.88 (s,1H, OH). **M.S.** (**M/z**):- 554(M+1).

Antimicrobial activity

Antimicrobial screening was used disc diffusion method(H. Afaf,2000) at а concentration of 100µg/ml, adopted with some modification for the prepared compound using Penciline and Streptomycin as references. The prepared compounds were tested against one strain of Gram +ve bacteria, Gram -ve The compounds were bacteria, fungi. evaluated for antibacterial activity against Staphylococcus aureus gr +ve, Escherichia coli gr –ve Bacillus subtilis gr +ve, Salmonella *typhi gr –ve* and antifungal activity against Aspergillus oryzoe, Aspergillus niger, DMSO



was used as solvent control. The results of antimicrobial data are summarized in **Table 2**.

The compounds show the moderate to good activity against bacteria and fungi.

Compounds	Gram positive bacterias		Gram negative bacterias		Fungus	
	Staph aureus	Bacillus subtilis	Escherichia coli	S. typhi	Aspergillus oryzoe	Aspergillus niger,
3a	-	-	-	-	-	-
3b	+	+	-	-	-	-
3c	-	-	-	-	-	-
3d	+	+	-	-	+	+
3e	+	+	-	+	+	+
3f	-	-	-	-	-	-
3g	-	+	-	-	+	+
3h	-	+	-	-	+	+
3i	+	+	-	-	+	+
3ј	-	-	-	-	+	+
3k	+	+	-	-	-	-
DMSO	-	-	-	-	+	-
Penicilline 1	+	+	+	+	X	х
Streptomycin 2	++	++	++	++	X	х
Greseofulvin	X	X	X	X	-	-

 Table-2. Antimicrobial activity of Chalcones (3a-k).

(++) = Clear Zone of Inhibition, (+) = Minimum Zone of Inhibition, (-) = No Effect, X = Not applicable, Standerd [1] Penciline (+), Standard [2] ,Streptomycin (++), Greseofulvin (fungus)

4. Conclusion

In this work, we have synthesized Chalcones (3a-k) by the reaction of 2-hydroxy acetophenone with substituted aldehydes with good yield, this method has the advantage of mild and efficient chemistry techniques, reaction time is short, the workout is easy. The newly synthesized Chalcones (3a-k) were

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confirmed by spectral analysis and further evaluated for their antimicrobial activity.

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